

Contents lists available at ScienceDirect

# Journal of Environmental Sciences



journal homepage: www.elsevier.com/locate/jes

# *In situ* nitrogen removal processes in intertidal wetlands of the Yangtze Estuary

Cheng Liu<sup>1</sup>, Lijun Hou<sup>1,\*</sup>, Min Liu<sup>2,3</sup>, Yanling Zheng<sup>1,2,3</sup>, Guoyu Yin<sup>2,3</sup>, Hongpo Dong<sup>1</sup>, Xia Liang<sup>1</sup>, Xiaofei Li<sup>4</sup>, Dengzhou Gao<sup>2,3</sup>, Zongxiao Zhang<sup>1</sup>

<sup>1</sup> State Key Laboratory of Estuarine and Coastal Research, East China Normal University, Shanghai 200241, China

<sup>2</sup> Key Laboratory of Geographic Information Science (Ministry of Education), East China Normal University, Shanghai 200241, China

<sup>3</sup> School of Geographic Sciences, East China Normal University, Shanghai 200241, China

<sup>4</sup> Key Laboratory for Humid Subtropical Eco-geographical Processes of the Ministry of Education, Fujian Normal University, Fuzhou 350007, China

# ARTICLE INFO

Article history: Received 2 February 2020 Revised 3 March 2020 Accepted 16 March 2020 Available online 9 April 2020

Keywords: Denitrification Anammox Sediment Intertidal wetland Yangtze estuary

# ABSTRACT

Estuarine and intertidal wetlands are important sites for nitrogen transformation and elimination. However, the factors controlling nitrogen removal processes remain largely uncertain in the highly dynamic environments. In this study, continuous-flow experiment combined with <sup>15</sup>N isotope pairing technique was used to investigate *in situ* rates of denitrification and anaerobic ammonium oxidation (anammox) and their coupling with nitrification in intertidal wetlands of the Yangtze Estuary. The measured rates varied from below the detection limit to 152.39 µmol N/(m<sup>2</sup>·hr) for denitrification and from below the detection limit to 43.06 µmol N/(m<sup>2</sup>·hr) for anammox. The coupling links of nitrogen removal processes with nitrification were mainly dependent on nitrate, organic carbon, sulfide, dissolved oxygen and ferric iron in the estuarine and intertidal wetlands. Additionally, it was estimated that the actual nitrogen removal processes annually removed approximately 5% of the terrigenous inorganic nitrogen discharged into the Yangtze Estuary. This study gives new insights into nitrogen transformation and fate in the estuarine and intertidal wetlands.

© 2020 Published by Elsevier B.V. on behalf of The Research Centre for Eco-Environmental Sciences, Chinese Academy of Sciences.

#### Introduction

Nitrogen (N) is a limiting nutrient for eutrophication in coastal marine ecosystems (Damashek and Francs, 2017; Howarth and Marino, 2006). With increased human activities (*e.g.*, excessive application of nitrogen fertilizers and fossil fuel burning), a large amount of reactive nitrogen has been released into aquatic ecosystems through riverine runoff, atmospheric deposition and sewage discharge during the past several decades (Finlay et al., 2013; Galloway et al., 1996; Galloway et al., 2003). Since the era of industrial revolution, the input of reactive nitrogen has been nearly doubled into estuarine and coastal regions (Galloway et al., 2004; Howarth et al., 2011). Excess N loading has thus led to many aquatic ecological issues, such as eutrophication, water quality deterioration and hypoxia (Yabe et al., 2009). Therefore, understanding nitrogen removal processes and their controlling

\* Corresponding author. E-mail address: ljhou@sklec.ecnu.edu.cn (L. Hou). factors is essential for the elimination of nitrogen pollution in estuarine and coastal environments.

Dissimilatory nitrate reduction processes, denitrification and anaerobic ammonium oxidation (anammox), can remove reactive nitrogen from aquatic ecosystems permanently through converting oxidized nitrogen  $(NO_x^-)$  to dinitrogen gas  $(N_2)$  (Yin et al., 2015). Therefore, these two processes play a vital role in counteracting the adverse effects caused by nitrogen overload. Depending on the source of NO<sub>x</sub><sup>-</sup>, nitrogen removal processes can generally be identified as uncoupled denitrification/anammox (Dw/Aw) which remove the  $NO_x^-$  from the overlying water, and coupled nitrification-denitrification/anammox  $(D_n/A_n)$  which eliminate the  $NO_x^{-}$  from the nitrification process (Jetten et al., 2001; Mulder et al., 1995; Nielsen, 1992; Risgaard-Petersen et al., 2003). The activity of uncoupled nitrogen removal processes (NR<sub>w</sub>) and coupled nitrification-nitrogen removal processes (NRn) are closely related to environmental factors. For instance, Seitzinger et al. (2006) indicated that high  $NO_x^{-}$  concentration in overlying water favors the uncoupled denitrification while low NO<sub>x</sub><sup>-</sup> concentration in overlying water is beneficial to the coupled nitrification-denitrification.

https://doi.org/10.1016/j.jes.2020.03.005

1001-0742/© 2020 Published by Elsevier B.V. on behalf of The Research Centre for Eco-Environmental Sciences, Chinese Academy of Sciences.

Previous studies suggested that under hypoxic condition,  $NR_n$  is inhibited, mainly due to the adverse effect of low dissolved oxygen (DO) on aerobic nitrification and its coupling with nitrogen removal processes (Kuparinen and Tuominen, 2001; Li et al., 2019; Rasmussen, 1992). However, low DO may be in favor of  $NR_n$  in some cases (Hansen and Blackburn, 1991; Rysgaard-Petersen et al., 1994). Under anoxic condition, the final form of nitrogen released from the sediment is altered, and ammonium generally dominates the efflux of nitrogen. With the increase of ammonium concentration, nitrification may be enhanced and thus stimulate  $NR_n$  (Abbasi and Adams, 2000; Enrich-Prast et al., 2016). Moreover, salinity variation can change the link between nitrification and nitrogen removal processes (Hu et al., 2019; Liu et al., 2019; She et al., 2016). However, the factors controlling  $NR_w$  and  $NR_n$  remain largely uncertain especially in a specific aquatic ecosystem.

Under the tidal fluctuation, the sediment in estuarine and intertidal wetlands experiences periodic exposure and inundation, which causes frequent redox changes (Robinson et al., 2006). Due to high sensitivity of nitrogen transformation to redox state, the intertidal area is thus a hotspot of nitrogen cycle (Damashek and Francis, 2017). During the ebb tide, oxic condition can enhance the activity of nitrification, and then during the flood tide, anoxic condition may support higher rates of NR<sub>n</sub> (Jiao et al., 2018; Senga et al., 2019). Therefore,  $NR_n$  may be the main nitrogen removal pathway in intertidal wetlands (Peng et al., 2016). However, the intertidal ecosystems have been subjected to excessive loading of artificial contaminants (e.g., reactive nitrogen and organic matter) in recent decades, which may alter the relative importance of NR<sub>n</sub> and NRw in intertidal wetlands (Galloway et al., 2004; Howarth et al., 2011; Liu et al., 2019; Peng et al., 2016). Therefore, it is of environmental and ecological significance to understanding the interaction of nitrogen removal processes with environmental factors and the relative importance of NR<sub>n</sub> and NR<sub>w</sub> for the intertidal wetlands.

Our previous research determined the potential rates of denitrification and anammox in the Yangtze estuarine and intertidal wetlands, using sediment-slurry incubation experiments (Hou et al., 2013). However, slurry incubation greatly changes nitrate availability and biogeochemical gradients, due to the mixing of the sediment. Therefore, the measured rates through sediment-slurry experiments cannot reflect the actual nitrogen removal from the sediment (Behrendt et al., 2013). In contrast, continuous-flow experiments could preserve the structural integrity of the sediments and simulate water flow, to some extent reflecting the in situ environment (Hou et al., 2012; Aoki and Mcglathery, 2017). In the present study, continuous-flow experiments combined with <sup>15</sup>N isotope pairing technique were thus conducted (1) to determine the actual rates of nitrogen removal processes and discriminate the uncoupled and coupled nitrogen removal processes, (2) to illustrate the main factors controlling the actual nitrogen removal rates and determining the relative importance of both NR<sub>n</sub> and NR<sub>w</sub>, and (3) to estimate environmental implications of nitrogen removal processes in intertidal wetlands of the Yangtze Estuary. This study helps us improve the understanding of nitrogen cycling in the estuarine and intertidal wetlands.

#### 1. Materials and methods

#### 1.1. Study area and sample collection

The Yangtze Estuary is the largest estuary in China, connecting the Yangtze River and the East China Sea. It is located in a typical subtropical monsoon region. Under the interaction between riverine runoff and tidal current, a marked salinity gradient occurs along the estuary. Extensive intertidal wetlands have developed along the estuarine and coastal zone, mainly due to the de-



**Fig. 1.** Sampling sites along the intertidal wetlands of the Yangtze Estuary. LCG: Luchaogang; DH: Donghai; BLG: Bailonggang; SDK: Shidongkou; LHK: Liuhekou; XP: Xupu.

position of tremendous discharge of suspended sediment from the catchment. In this study, six sampling sites were selected along the intertidal wetlands of the Yangtze Estuary (Fig. 1), including Luchaogang (LCG), Donghai (DH), Bailonggang (BLG), Shidongkou (SDK), Liuhekou (LHK) and Xupu (XP). Sites BLG, SDK, LHK and XP are located in the freshwater area, while sites LCG and DH are located in the brackish area. Field surveys were conducted in April (spring), July (summer), October (autumn) 2017, and January (winter) 2018. At each sampling site, six intact cores (7.0 cm in diameter and 15 cm deep) were collected during the ebb period. Three of these cores were used to carry out the continuousflow experiments. Surface sediment (0-5 cm deep) was also taken from the remaining cores for sediment characteristics measurements and slurry incubation experiments. Meanwhile, 30 L of overlying tidal water was collected from each site for continuous-flow experiments.

# 1.2. Continuous-flow experiments

Three replicate intact sediment cores from each site were installed respectively into a gas-tight continuous-flow system (Yin et al., 2016; Liu et al., 2019). The system was mainly composed of a gas-tight water bag (Tedlar), an acetol plunger with a Viton o-ring and two Peek tubings, an intact sediment core, and a multi-channel peristaltic pump (Appendix A Fig. S1). The acetol plunger was placed about 5 cm above the water-sediment interface. One end of inlet Peek tubing was near the sediment surface, and the other end was connected through a peristaltic pump with a gas-tight bag containing overlying tidal water enriched with <sup>15</sup>N- $NO_3^-$  (90–99% <sup>15</sup>N-KNO<sub>3</sub>) to final concentrations of 100  $\mu$ mol/L <sup>15</sup>N-NO<sub>3</sub><sup>-</sup> (depending on *in situ* nitrate concentrations). One end of outlet Peek tubing was near the interface between the acetol plunger and overlying tidal water, and the other end was used for collecting incubation water. The water in the incubation system was replaced at a rate of 1.5 mL/min (Gardner and Mccarthy, 2009). The continuous-flow experiments were conducted at near in situ temperature and dissolved oxygen. After an overnight pre-incubation, the continuous-flow system reached a steady state (Mctigue et al., 2016). Within the next 24 hr, inlet and outlet water samples were collected with gas-tight vials (12 mL Exetainer, Labco, UK) every 6 hr. After collection, 200 µL saturated ZnCl<sub>2</sub> solution was injected into each vial to eliminate microbial activity. The concentrations of dissolved nitrogen gasses (including  $^{29}N_2$  and  $^{30}N_2$ ) in water samples were measured by membrane inlet mass spectrometry (HPR-40, Hiden Analytical, UK), with a detection limit of 0.1 µmol N<sub>2</sub>/L (Yin et al., 2014). The actual denitrification (D<sub>14</sub>) and anammox (A<sub>14</sub>) rates, as well as the uncoupled



**Fig. 2.** Actual rates of (a) denitrification  $(D_{14})$  and (b) anammox  $(A_{14})$  in the intertidal wetlands of the Yangtze Estuary.

 $(D_w \text{ and } A_w)$  and coupled  $(D_n \text{ and } A_n)$  denitrification/anammox rates were estimated according to the concentration changes of dissolved nitrogen gasses during the continuous-flow incubation (Liu et al., 2019). A detailed calculation method is provided in Appendix A.

#### 1.3. Analysis of environmental factors

The in situ temperature, salinity, pH and DO in overlying tidal water were measured using a multi-parameter water quality meter (AZ86031, AZ Instrument, China). Concentrations of dissolved inorganic nitrogen (including  $NH_4^+$ ,  $NO_3^-$  and  $NO_2^-$ ) in overlying tidal water were analyzed by a continuous-flow nutrient analyzer (SAN Plus, Skalar Analytical B.V., the Netherlands), with detection limits of 0.5  $\mu$ mol/L for NH<sub>4</sub><sup>+</sup> and 0.1  $\mu$ mol/L for NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> (Hou et al., 2015). Water content of sediment (WC) was calculated from weight loss of fresh sediment by freeze-drying. Grain size of sediment was determined using a LS 13 320 Laser Grain Sizer (Beckman Coulter, USA). Sediment organic carbon (OC) was analyzed by the K<sub>2</sub>CrO<sub>7</sub>-FeSO<sub>4</sub> method (Wang et al., 2007). Sediment sulfide concentration was determined using Sure-Flow combination silversulfide electrode (Orion, USA), with a detection limit of 0.1 µmol/L (Hou et al., 2012). Total extractable ferrous iron and ferric iron in sediments were analyzed using the ferrozine method described by Lovley et al. (1987). Sediment inorganic nitrogen was extracted using a 2 mol/L KCl solution and then analyzed using a continuousflow nutrient analyzer (Liu et al., 2019). The data on these measured environmental factors are provided in Appendix A Table S1 and Table S2.

#### 1.4. Data analysis

Analysis of variance (ANOVA) was used to examine the spatial and temporal changes in nitrogen removal processes (including  $D_{14}$ ,  $A_{14}$ ,  $D_w$ ,  $A_w$ ,  $D_n$ , and  $A_n$ ). Pearson's correlation was performed to analyze the links of nitrogen removal processes with environmental variables. Redundancy analysis (RDA) was conducted to illustrate the effects of environmental factors on the uncoupled ( $D_w$  and  $A_w$ ) and coupled ( $D_n$  and  $A_n$ ) nitrogen removal processes, using CANOCO 5.

#### 2. Results

# 2.1. Actual rates of denitrification and anammox

The rates of  $D_{14}$  and  $A_{14}$  varied from below the detection limit to 152.39 µmol N/(m<sup>2</sup>·hr) and from below the detection limit to 43.06 µmol N/(m<sup>2</sup>·hr), respectively (Fig. 2). In the study area, denitrification was the dominant nitrogen removal process, contributing to about 89% of the total nitrogen removal (sum of denitrification and anammox). A significant spatial difference was



**Fig. 3.** Correlations of actual nitrogen removal rates with physicochemical characteristics of (a) overlying tidal water and (b) sediment. TEMP: temperature; SAL: salinity; DO: dissolved oxygen; wNO<sub>3</sub><sup>-</sup>: NO<sub>3</sub><sup>-</sup> concentration in overlying water; wNO<sub>2</sub><sup>-</sup>: NO<sub>2</sub><sup>-</sup> concentration in overlying water; wHu<sub>4</sub><sup>+</sup>: NH<sub>4</sub><sup>+</sup> concentration in overlying water; WC: water content; M\Phi: median grain size; OC: organic carbon; Fe<sup>2+</sup>: ferrous iron; Fe<sup>3+</sup>: ferric iron; SNO<sub>3</sub><sup>-</sup>: NO<sub>3</sub><sup>-</sup> concentration in sediment; sNU<sub>2</sub><sup>-</sup>: NO<sub>2</sub><sup>-</sup> concentration in sediment; sNH<sub>4</sub><sup>+</sup>: NH<sub>4</sub><sup>+</sup> concentration in sediment.

found in the D<sub>14</sub> rates (two-factor ANOVA, F = 218.813, df = 5, P < 0.001) and A<sub>14</sub> rates (two-factor ANOVA, F = 643.867, df = 5, P < 0.001). In general, these two nitrogen removal rates were higher at the freshwater sites than at the brackish sites. There were no identical seasonal changes for D<sub>14</sub> and A<sub>14</sub> rates among all sampling sites. According to annual data of nitrogen removal rates and environmental factors, we conducted Pearson's correlation analysis (Fig. 3). The results showed that both D<sub>14</sub> and A<sub>14</sub> rates were not only positively correlated with nitrite (r = 0.58, P < 0.01 and r = 0.65, P < 0.01, respectively) in overlying tidal water but also positively correlated with OC (r = 0.60, P < 0.01 and r = 0.53, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrite (r = 0.85, P < 0.01 and r = 0.51, P < 0.01, respectively) and nitrie (

i Jan. 2018

Nov. 2017

DH BLG SDK LHK

b

LĊG

LCG DH BLG SDK

<sup>3</sup>]d

(µmol N/(m<sup>2</sup>·hr))

N/(m<sup>2</sup>·hr))

х́р

**Fig. 4.** Rates of (a) uncoupled denitrification  $(D_w)$ , (b) coupled nitrificationdenitrification  $(D_n)$ , (c) uncoupled anammox  $(A_w)$ , and (d) coupled nitrificationanammox  $(A_n)$  in the intertidal wetlands of the Yangtze Estuary.

ХP

///// Jul. 2017

0.05, respectively) in sediment. In addition, the  $D_{14}$  rates were also positively correlated with sulfide (r = 0.51, P < 0.05) in sediment.

## 2.2. Uncoupled/coupled nitrogen removal process

NR<sub>w</sub> includes D<sub>w</sub> and A<sub>w</sub>, and NR<sub>n</sub> includes D<sub>n</sub> and A<sub>n</sub>. In this study, the rates of D<sub>w</sub> and A<sub>w</sub> ranged from below the detection limit to 152.39 µmol N/(m<sup>2</sup>·hr) and from below the detection limit to 43.06 µmol N/(m<sup>2</sup>·hr), respectively, while the rates of D<sub>n</sub> and A<sub>n</sub> varied from below the detection limit to 20.87 µmol N/(m<sup>2</sup>·hr) and from below the detection limit to 20.40 µmol N/(m<sup>2</sup>·hr), respectively (Fig. 4). The average rate of NR<sub>w</sub> (31.12 µmol N/(m<sup>2</sup>·hr)) was much higher than that of NR<sub>n</sub> (4.48 µmol N/(m<sup>2</sup>·hr)), indicating that NR<sub>w</sub> was the main nitrogen removal pathway in intertidal wetlands of the Yangtze Estuary. The rates of NR<sub>w</sub> were generally higher at the freshwater sites than at the brackish sites, while the rates of NR<sub>n</sub> increased with the increasing salinity.

#### 2.3. Effects of environmental factors on nitrogen removal pathways

RDA was used to investigate the effects of environmental factors on uncoupled and coupled nitrogen removal processes. According to the annual data of different nitrogen removal rates and environmental factors, we performed the RDA analysis (Fig. 5). The first two RDA axes (principal components) explained 66.9% of the cumulative percentage variance. The first and second axes accounted for 48.5% and 18.4% of the total variations of D<sub>w</sub>, D<sub>n</sub>, A<sub>w</sub>, and A<sub>n</sub>, respectively. NR<sub>w</sub> rates were positively correlated with contents of inorganic nitrogen, OC and sulfide. NR<sub>n</sub> rates were positively correlated with contents of WC and ferric iron. It is shown that the environmental factors played an important role in regulating the rates of different nitrogen removal pathways.

#### 3. Discussion

Due to human activities, the nitrogen load imported into aquatic ecosystems has greatly increased, and led to severe nitrogen pollution. At present, denitrification and anammox are considered to be the two main pathways of N removal, which can effectively relieve the adverse effects caused by nitrogen overload (Yin et al., 2015). Although estuarine wetlands occupy only a small area of the earth's land surface, they play an important role in global nitrogen removal (Mctigue et al., 2016; Seldomridge and Prestegaard,



Fig. 5. RDA analyses of uncoupled and coupled nitrification nitrogen removal processes with environment factors.

2011; Zheng et al., 2019). Our previous study determined the potential rates of denitrification and anammox in intertidal wetlands of the Yangtze Estuary through sediment slurry incubations (Hou et al., 2013), which generally overestimated the ability of sediment nitrogen removal. In order to obtain the actual rates of denitrification and anammox in the study area, continuous-flow experiments were conducted in combination with <sup>15</sup>N isotope pairing technique. According to our results, the rates of D<sub>14</sub> and A<sub>14</sub> ranged from below the detection limit to 152.39  $\mu$ mol N/(m<sup>2</sup>·hr) and from below the detection limit to 43.06  $\mu$ mol N/(m<sup>2</sup>·hr), respectively. Both of D<sub>14</sub> rates and A<sub>14</sub> rates were generally higher in the freshwater area than in the brackish area. Among the 6 sampling sites, the highest D<sub>14</sub> and A<sub>14</sub> rates appeared at site SDK, which might be related to sewage discharge near the site. Sewage containing a lot of nitrogen was conducive to producing more organic matter and sulfide and forming anoxia condition, which was in favor of nitrogen removal processes (Senga et al., 2019). However, although BLG site was also near a sewage outlet, the  $D_{14}$  and  $A_{14}$  rates at this site were lower, which was likely attributed to sediment grain size. The grain size of sediment at BLG was larger than that of other sampling sites (Appendix A Table S2). In general, coarse sediment with small specific surface area is not favorable to the attachment of nitrogen removal microorganisms and thus decreases the rate of nitrogen removal process (Xia et al., 2017). The actual nitrogen removal rates in intertidal wetlands of the Yangtze Estuary were several times higher than those examined in the adjacent sea of the Yangtze Estuary (Liu et al., 2019). There may be two reasons for this difference. Firstly, compared with the adjacent coastal sea, the intertidal wetland has higher nitrogen load in overlying water, which is beneficial to nitrogen removal process. Secondly, during the ebb tide, air enters the intertidal wetland, and the substrate is replenished with oxygen, subsequently promoting nitrification to produce more nitrate/nitrite for nitrogen removal processes (Li et al., 2015a).

Environmental factors play an important role in controlling nitrogen removal. However, for a specific environment, the environmental factors controlling nitrogen removal processes remain largely uncertain (Enrich-Prast et al., 2016). In the present study, the D<sub>14</sub> rates were positively related to nitrite (r = 0.58, P < 0.01) in the overlying tidal water and nitrite (r = 0.80, P < 0.01) and nitrate (r = 0.47, P < 0.05) in sediment, indicating that the importance of NO<sub>x</sub><sup>-</sup> as the substrate in controlling denitrification. This result was in consistence with other studies (Koop-Jakobsen

<sup>180</sup>] a

D<sub>w</sub> (µmol N/(m²·hr)) 00 07 00

₫0

50

((.14; m)/N lount) \*

₹ 10

LĊG

С

LCG DH

Apr. 2017

16 B B

DH BLG SDK LHK

BLG SDK LHK

and Giblin, 2010; Liu et al., 2018). Canonical denitrification belongs to heterotrophic dissimilatory nitrate reduction, in which organic matter is required to provide energy and electrons (Gao et al., 2017). Therefore, more organic matter is conducive to denitrification, which was confirmed by the positive relationship between the  $D_{14}$  rates and OC in our study (Fig. 3). Sulfide is a toxic substance, and high concentration of sulfide can threaten aquatic ecosystems. Some studies reported that the excessive sulfide in sediment can reduce the denitrification rate by inhibiting the metabolism of denitrifiers (Aelion and Warttinger, 2010; Brunet and Garcia-Gil, 1996). However, it was also found that sulfide as a strong reducing substance can promote denitrification by providing electrons (Deng et al., 2015; Yang et al., 2015). This contradiction might depend on the form of sulfide. H<sub>2</sub>S can inhibit denitrification, while FeS and elemental sulfur may be in favor of denitrification (Burgin and Hamilton, 2007). In the present study, there was a significant positive correlation between sulfide and ferrous iron (Fig. 3), suggesting that FeS may be the main sulfide form in the Yangtze estuarine and intertidal wetlands. It explained why there was a positive correlation between the D<sub>14</sub> rates and sulfide.

Ammonium and nitrite are the substrates of anammox reaction, so anammox activity may depend on the levels of ammonium and nitrite. In this study, it was observed that the A<sub>14</sub> rates were not correlated to ammonium but to nitrite in the overlying water (r = 0.65, P < 0.01) and sediment (r = 0.51, P < 0.05) (Fig. 3). These relationships indicated that nitrite is the limiting substrate for anammox activity in intertidal wetlands of the Yangtze Estuary. It's worth noting that the autotrophic anammox process was positively correlated with organic carbon (r = 0.60, P < 0.01). Previous studies have reported that nitrite produced by denitrification was a significant source for anammox (Gao et al., 2017; Shan et al., 2016), which was supported by the significant correlation between the A<sub>14</sub> and D<sub>14</sub> rates in our study (r = 0.84, P < 0.01). Therefore, organic carbon might indirectly promote the rate of A<sub>14</sub> through stimulating denitrification activity.

According to the sources of  $NO_x^-$ , nitrogen removal processes can be divided into  $NR_w$  in which  $NO_x^-$  is derived from the overlying water and  $NR_n$  in which  $NO_x^-$  is obtained from nitrification (Fernandes et al., 2016; Nielsen, 1992; Risgaard-Petersen et al., 2003). Previous studies show that the rates of  $NR_w$  were high under aquatic ecosystems with high nitrate concentrations, organic matter contents, or reducing substances (e.g., sulfide and ferrous iron), while the rates of NR<sub>n</sub> were high under aquatic ecosystems with high concentrations of ammonium or DO (Abbasi and Adams, 2000; Burgin and Hamilton, 2007; Liu et al., 2019; Mctigue et al., 2016; Neubacher et al., 2011; Seitzinger et al., 2006;). So, the changes of NR<sub>w</sub> and NR<sub>n</sub> rates strongly depend on the environmental factors of aquatic ecosystems. Due to human activity,  $NO_x^$ concentration in tidal water of the Yangtze Estuary has greatly increased in recent decades. In the 1960s, the concentration of  $NO_x^-$  in tidal water was below 10 µmol/L, but now it increased to about 120 µmol/L (Yang et al., 2019). During the flood tide, the intertidal sediment is submerged by tidal water with high concentration of nitrate. High concentration of nitrate is beneficial to the diffusion of nitrate into the sediment (Chen et al., 2015), thus enhancing the availability of nitrate to dissimilatory nitrate reducers in the sediment (Yin et al., 2015). In this case, the rates of NRw were greatly promoted in intertidal wetlands of the Yangtze Estuary (Fig. 5), which was consistent with previous studies (Koop-Jakobsen and Giblin, 2010; Liu et al., 2019; Piehler and Smyth, 2011; Seitzinger et al., 2006). On the other hand, high nitrate concentration might also be in favor of the production of organic matter and sulfide (Senga et al., 2019). This assumption was supported by the positive correlations of inorganic nitrogen with organic carbon and sulfide in sediment (Fig. 3). Because organic matter and sulfide can directly provide energy and electrons or indirectly



Fig. 6. Percentage of uncoupled nitrogen removal processes  $(NR_w)$  and coupled nitrification-nitrogen removal processes  $(NR_n)$  in the intertidal wetlands of the Yangtze Estuary.

supply more  $NO_2^-$  for  $NR_w$ , the activity of  $NR_w$  was enhanced with the increase of organic carbon and sulfide (Fig. 5). However, there was no correlation between  $NR_n$  and organic carbon (or sulfide). The main pathway of NR<sub>n</sub> includes two steps: ammonia oxidation (AO) and nitrite reduction (NIR), and AO is the first step of nitrification which is an autotrophic process and a rate-limiting step of NR<sub>n</sub> (Miao et al., 2018; Liu et al., 2019). Therefore, the organic carbon and sulfide had a little effect on AO and NR<sub>n</sub>. Compared with summer, the remaining seasons had high rates of  $NR_n$  (Fig. 4b and d). This was likely because the DO concentrations were higher in other seasons than in summer (Appendix A Table S1), which was in favor of nitrification. An interesting result in this study was that with the increase of ferric iron concentrations, the rates of NR<sub>n</sub> were promoted (Fig. 5). According to Li et al. (2015b), ferric iron could promote the rates of anaerobic ammonium oxidation coupled with Fe(III) reduction (Feammox), which produces more  $NO_x^-$  and in turn favors  $NR_n$  (Appendix A Fig. S2).

Previous studies demonstrated that NRn was an important nitrogen removal pathway in estuarine and intertidal wetlands (Jenkins and Kemp, 1984; Peng et al., 2016; Seitzinger et al., 2006; Welsh et al., 2000; White and Howes, 1994). This was partly because tide provided sufficient oxygen for nitrification to produce a large amount of  $NO_x^{-}$  and subsequently the nitrification-derived  $NO_x^-$  fueled nitrogen removal processes (An and Joye, 2001; Senga et al., 2019). However, this study found that compared with NR<sub>n</sub>, NR<sub>w</sub> was the dominant nitrogen removal pathway, contributing to about 89% of total nitrogen removal in intertidal wetlands of the Yangtze Estuary. The Yangtze Estuary had high  $NO_x^-$  concentrations in tidal water due to intensive human activities, which may alter the relative importance of both NR<sub>n</sub> and NR<sub>w</sub>. In the intertidal wetlands with low  $NO_x^{-}$  concentrations, nitrogen removal microorganisms can primarily rely on  $NO_x^-$  derived from the product of nitrification and oxygen is the dominant factor indirectly controlling nitrogen removal. In contrast, in the intertidal wetlands with high  $NO_x^-$  concentrations, nitrogen removal microorganisms could obtain sufficient NO<sub>x</sub><sup>-</sup> from overlying water, and NR<sub>w</sub> was thus a primary nitrogen removal pathway (Piehler and Smyth, 2011; Seitzinger et al., 2006). Although NR<sub>n</sub> was not dominated, the relative importance of NRn also showed the significant temporal and spatial variations in intertidal wetlands of the Yangtze Estuary (Fig. 6). The relative importance of NR<sub>n</sub> in summer was generally lower than that in other seasons, which was likely attributed to low DO in summer. Low DO is adverse to aerobic nitrification and NR<sub>n</sub>, but is beneficial to anaerobic NR<sub>w</sub> (Rasmussen, 1992). The relative importance of NR<sub>n</sub> was generally higher at the brackish sites than at the freshwater sampling sites. There are two reasons for this spatial difference. Firstly, due to the dilution of sea water, the concentration of nitrate in the brackish area was lower than that in the freshwater area (Appendix A Table S1), which thus enhanced the relative importance of NR<sub>n</sub>. Secondly, salinity might play an important role in controlling the relative importance of different nitrogen removal pathways. Some studies suggested that as the salinity increases, ammonia oxidation is promoted, while nitrite oxidation is inhibited, which leads to the accumulation of NO<sub>2</sub><sup>-</sup> (She et al., 2016). A large amount of NO<sub>2</sub><sup>-</sup> accelerates nitrite reduction, and the relative importance of NR<sub>n</sub> might thus be enhanced with the increase of salinity.

Nitrogen removal processes of intertidal wetland are critical to reduce nitrogen overload in estuarine and coastal ecosystems. Based on the average of actual nitrogen removal rates in the study area, the annual amount of removed nitrogen was estimated according to the following equation (Lin et al., 2017):

 $F = R \times a \times s \times t$ 

where F(t N) denotes the annual amount of nitrogen removal; R( $\mu$ mol N/( $m^2 \cdot hr$ )) denotes the annual average of nitrogen removal rate in the study area; a denotes the unit conversion factor (0.14); s $(km^2)$  denotes the area (approximately  $1.2 \times 10^4 km^2$ ) of the tidal flats and adjacent zone along the Yangtze Estuary; and t (year) denotes the time. It is estimated that approximately  $5.2 \times 10^4$  t N was annually removed from the study area, which accounted for approximately 5% of the total terrigenous inorganic nitrogen discharged into the Yangtze Estuary (Lin et al., 2017). The calculated contribution to nitrogen removal was lower than the estimation (25%) from sediment-slurry incubation (Deng et al., 2015). This comparison partly reflects that the role of sediment in nitrogen removal is largely overestimated by sediment-slurry experiments. Therefore, the actual capacity of nitrogen removal should be taken into account for the control of nitrogen pollution in the estuarine and intertidal wetlands.

#### 4. Conclusions

This study first quantified the in situ rates of nitrogen removal processes, denitrification and anammox, and their coupling links with nitrification in the Yangtze estuarine and intertidal wetlands. The in situ rates of denitrification and anammox varied from below the detection limit to 152.39  $\mu$ mol N/(m<sup>2</sup>·hr) and from below the detection limit to 43.06  $\mu$ mol N/(m<sup>2</sup>·hr), respectively. Meanwhile, the rates ranged from below the detection limit to 195.45 µmol  $N/(m^2 \cdot hr)$  for the uncoupled nitrogen removal processes and from below the detection limit to 21.35  $\mu$ mol N/(m<sup>2</sup>·hr) for the coupled nitrogen removal processes. The uncoupled nitrogen removal pathways were generally associated with nitrate, OC and sulfide, while the coupled nitrogen removal pathways were related to DO and ferric iron. It was estimated that in situ denitrification and anammox processes annually removed approximately  $5.2 \times 10^4$  t N from the study area, which was approximate to 5% of the total terrigenous inorganic nitrogen discharged into the Yangtze estuarine and coastal environments.

#### Acknowledgments

This work was supported by the National Natural Science Foundation of China (Nos. 41725002, 41671463, 41601530, 41761144062, and 41730646) and the Fundamental Research Funds for the Central Universities and Chinese National Key Programs for Fundamental Research and Development (Nos. 2016YFA0600904, 2016YFE0133700). Data presented here can be obtained by sending a request to the corresponding author.

## Appendix A. Supplementary materials

Supplementary material associated with this article can be found, in the online version, at doi:10.1016/j.jes.2020.03.005.

# References

- Abbasi, M.K., Adams, W.A., 2000. Estimation of simultaneous nitrification and denitrification in grassland soil associated with urea-N using <sup>15</sup>N and nitrification inhibitor. Biol. Fert. Soils. 31, 38–44.
- Aelion, C.M., Warttinger, U., 2010. Sulfide inhibition of nitrate removal in coastal sediments. Estuar. Coast. 33, 798–803.
- An, S., Joye, S., 2001. Enhancement of coupled nitrification-denitrification by benthic photosynthesis in shallow estuarine sediments. Limnol. Oceanogr. 46, 62–74.
- Aoki, L.R., Mcglathery, K.J., 2017. Push-pull incubation method reveals the importance of denitrification and dissimilatory nitrate reduction to ammonium in seagrass root zone. Limnol. Oceanogr-meth. 15, 766–781.
- Behrendt, A., Beer, D.D., Stief, P., 2013. Vertical activity distribution of dissimilatory nitrate reduction in coastal marine sediments. Biogeosciences 10, 7509–7523.
- Brunet, R.C., Garcia-Gil, L.J., 1996. Sulfide-induced dissimilatory nitrate reduction to ammonia in anaerobic freshwater sediments. Fems. Microbiol. Ecol. 21, 131–138.
- Burgin, A.J., Hamilton, S.K., 2007. Have we overemphasized the role of denitrification in aquatic ecosystems? A review of nitrate removal pathways. Front. Ecol. Environ. 5, 89–96.
- Cheng, X.L., Hou, L.J., Liu, M., Zheng, Y.L., Yin, G.Y., Li, X.F., et al., 2015. Inorganic nitrogen exchange across the sediment–water interface in the eastern Chongming tidal flat of the Yangtze Estuary. Environ. Earth. Sci. 74, 2173–2184.
- Deng, F.Y., Hou, L.J., Liu, M., Zheng, Y.L., Yin, G.Y., Li, X.F., et al., 2015. Dissimilatory nitrate reduction processes and associated contribution to nitrogen removal in sediments of the Yangtze Estuary. J. Geophys. Res-biogeo. 120, 1521–1531.
- Enrich-Prast, A., Lúcia Santoro, A.S., Coutinho, R., Nielsen, L.P.A., Esteves, F., 2016. Sediment denitrification in two contrasting tropical shallow lagoons. Estuar. Coast. 39, 657–663.
- Fernandes, S.O., Javanaud, C., Michotey, V.D., Guasco, S., Anschutz, P., Bonin, P., 2016. Coupling of bacterial nitrification with denitrification and anammox supports N removal in intertidal sediments (Arcachon Bay, France). Estuar. Coast. Shelf. S. 179, 39–50.
- Finlay, J.C., Small, G.E., Sterner, R.W., 2013. Human influences on nitrogen removal in lakes. Science 342, 247–250.
- Galloway, J.N., Aber, J.D., Erisman, J.W., Seitzinger, S.P., Howarth, R.W., Cowling, E.B., et al., 2003. The nitrogen cascade. Bioscience 53, 341–356.
- Galloway, J.N., Dentener, F.J., Capone, D.G., Boyer, E.W., Howarth, R.W., Seitzinger, S.P., et al., 2004. Nitrogen cycles: past, present, and future. Biogeochemistry 70, 153–226.
- Galloway, J.N., Howarth, R.W., Michaels, A.F., Nixon, S.W., Prospero, J.M., Dentener, F.J., 1996. Nitrogen and phosphorus budgets of the North Atlantic ocean and its watershed. Biogeochemistry 35, 3–25.
- Gao, D.Z., Li, X.F., Lin, X.B., Wu, D.M., Jin, B.S., Huang, Y.P., et al., 2017. Soil dissimilatory nitrate reduction processes in the Spartina alterniflora invasion chronosequences of a coastal wetland of southeastern China: dynamics and environmental implications. Plant. Soil. 421, 383–399.
- Gardner, W.S., Mccarthy, M.J., 2009. Nitrogen dynamics at the sediment-water interface in shallow, sub-tropical Florida Bay: why denitrification efficiency may decrease with increased eutrophication. Biogeochemistry 95, 185–198.
- Hansen, L.S., Blackburn, T.H., 1991. Aerobic and anaerobic mineralization of organic material in marine sediment microcosm. Mar. Ecol. Prog. Ser. 75, 283–291.
- Hou, LJ., Liu, M., Carini, S.A., Gardner, W.S., 2012. Transformation and fate of nitrate near the sediment-water interface of Copano Bay. Cont. Shelf. Res. 35, 86–94.
- Hou, LJ., Zheng, Y.L., Liu, M., Gong, J., Zhang, X.L., Yin, G.Y., et al., 2013. Anaerobic ammonium oxidation (anammox) bacterial diversity, abundance, and activity in marsh sediments of the Yangtze Estuary. J. Geophys. Res-biogeo. 118, 1237–1246.
- Hou, L.J., Zheng, Y.L., Liu, M., Li, X.F., Lin, X.B., Yin, G.Y., et al., 2015. Anaerobic ammonium oxidation and its contribution to nitrogen removal in China's coastal wetlands. Sci. Rep. 5, 15621.
- Howarth, R., Chan, F., Conley, D.J., Garnier, J., Doney, S.C., Marino, R., et al., 2011. . Coupled biogeochemical cycles: eutrophication and hypoxia in temperate estuaries and coastal marine ecosystems. Front. Ecol. Environ. 9, 18–26.
- Howarth, R., Marino, R., 2006. Nitrogen as the limiting nutrient for eutrophication in coastal marine ecosystems: evolving views over three decades. Limnol. Oceanogr. 51 (1 part 2), 364–376.
- Hu, M., Peñuelas, J., Sardans, J., Huang, J., Tong, C., 2019. Effects of nitrogen loading on emission of carbon gases from estuarine tidal marshes with varying salinity. Sci. Total. Environ. 667, 648–657.
- Jenkins, M., Kemp, W., 1984. The coupling of nitrification and denitrification in two estuarine sediments. Limnol. Oceanogr. 29, 609–619.
- Jetten, M.S., Wagner, M., Fuerst, J., Van, M.L., Kuenen, G., Strous, M., 2001. Microbiology and application of the anaerobic ammonium oxidation ('anammox') process. Curr. Opin. Biotech. 12, 283–288.
- Jiao, L., Wu, J., He, X., Wen, X., Li, Y., Hong, Y., 2018. Significant microbial nitrogen loss from denitrification and anammox in the land-sea interface of low permeable sediments. Int. Biodeter. Biodegr. 135, 80–89.

- Koop-jakobsen, K., Giblin, A.E., 2010. The effect of increased nitrate loading on nitrate reduction via denitrification and DNRA in salt marsh sediments. Limnol. Oceanogr. 55, 789–802.
- Kuparinen, J., Tuominen, L., 2001. Eutrophication and self-purification: counteractions forced by large-scale cycles and hydrodynamic processes. Ambio 30, 190–194.
- Li, J., Hu, Z., Li, F., Fan, J., Zhang, J., Li, F., et al., 2019. Effect of oxygen supply strategy on nitrogen removal of biochar-based vertical subsurface flow constructed wetland: Intermittent aeration and tidal flow. Chemosphere 223, 366–374.
- Li, L., He, C., Ji, G., Wei, Z., Sheng, L., 2015a. Nitrogen removal pathways in a tidal flow constructed wetland under flooded time constraints. Ecol. Eng. 81, 266–271.
- Li, X.F., Hou, L.J., Liu, M., Zheng, Y.L., Yin, G.Y., Lin, X.B., et al., 2015b. Evidence of nitrogen loss from anaerobic ammonium oxidation coupled with ferric iron reduction in an intertidal wetland. Environ. Sci. Technol. 49, 11560–11568.
- Lin, X.B., Liu, M., Hou, L.J., Gao, D.Z., Li, X.F., Lu, K.J., et al., 2017. Nitrogen losses in sediments of the East China Sea: Spatiotemporal variations, controlling factors and environmental implications. J. Geophys. Res-biogeo. 122, 2699–2715.
- Liu, C., Hou, L.J., Liu, M., Zheng, Y.L., Yin, G.Y., Han, P., et al., 2019. Coupling of denitrification and anaerobic ammonium oxidation with nitrification in sediments of the Yangtze Estuary: Importance and controlling factors. Estuar. Coast. Shelf. S. 220, 64–72.
- Liu, W.Z., Yao, L., Jiang, X.L., Guo, L.D., Cheng, X.L., Liu, G.H., 2018. Sediment denitrification in Yangtze lakes is mainly influenced by environmental conditions but not biological communities. Sci. Total. Environ. 616, 978–987.
- Lovley, D.R., Phillips, E.J.P., 1987. Rapid assay for microbially reducible ferric iron in aquatic sediments. Appl. Microbiol. Biot. 53, 1536–1540.
- Mctigue, N.D., Gardner, W.S., Dunton, K.H., Hardison, A.K., 2016. Biotic and abiotic controls on co-occurring nitrogen cycling processes in shallow Arctic shelf sediments. Nat. Commun. 7, 13145.
- Miao, Y.Y., Peng, Y.Z., Zhang, L., Li, B.K., Li, X.Y., Wu, L., et al., 2018. Partial nitrification-anammox (PNA) treating sewage with intermittent aeration mode: effect of influent C/N ratios. Chem. Eng. J. 334, 664–672.
- Mulder, A., Graaf, A.A.V.D., Robertson, L.A., Kuenen, J.G., 1995. Anaerobic ammonium oxidation discovered in a denitrifying fluidized bed reactor. Fems. Microbiol. Ecol. 16, 177–184.
- Neubacher, E.C., Parker, R.E., Trimmer, M., 2011. Short-term hypoxia alters the balance of nitrogen cycle in coastal sediments. Limnol. Oceanogr. 56, 651–665.
- Nielsen, L.P., 1992. Denitrification in sediment determined from nitrogen isotope pairing, Fems. Microbiol. Lett. 86, 357–362.
- Peng, X., Ji, Q., Angell, J.H., Kearns, P.J., Yang, H.J., Bowen, J.L., et al., 2016. . Long-term fertilization alters the relative importance of nitrate reduction pathways in salt marsh sediments. J. Geophys. Res. 121, 2082–2095.
- Piehler, M.F., Smyth, A.R., 2011. Habitat-specific distinctions in estuarine denitrification affect both ecosystem function and services. Ecosphere 2 (1 art 12), 1–16. Rasmussen, H., 1992. Microelectrode studies of seasonal oxygen uptake in a coastal
- sediment, role of molecular diffusion. Mar. Ecol. Prog. Ser. 81, 289–303. Risgaard-Petersen, N., Nielsen, L.P., Rysgaard, S., Dalsgaard, T., Meyer, R.L., 2003. Er-
- ratum: Application of the isotope pairing technique in sediments where anammox and denitrification co-exist. Limnol. Oceanogr-meth. 1, 63–73.

- Risgaard-Petersen, N., Rysgaard, S., Nielsen, L.P., Revsbech, N., 1994. Diurnal-variation of denitrification and nitrification in sediments colonized by benthic microphytes. Limnol. Oceanogr. 39, 573–579.
- Robinson, C., Gibbes, B., Li, L., 2006. Driving mechanisms for groundwater flow and salt transport in a subterranean estuary. Geophys. Res. Lett. 33, 155–170.
- Seitzinger, S., Harrison, J.A., Böhlke, J.K., Bouwman, A.F., Lowrance, R., Peterson, B., et al., 2006. Denitrification across landscapes and waterscapes: a synthesis. Ecol. Appl. 16, 2064–2090.
- Seldomridge, E., Prestegaard, K., 2011. Is denitrification kinetically-limited or transport-limited in tidal freshwater marshes? Appl. Geochem. 26, S256–S258.
- Senga, Y., Sato, T., Kuroiwa, M., Nohara, S., Suwa, Y., 2019. Anammox and denitrification in the intertidal sediment of the hypereutrophic Yatsu tidal flat. Japan Estuar. Coast. 42, 665–674.
- Shan, J., Zhao, X., Sheng, R., Xia, Y., Ti, C., Quan, X., et al., 2016. Dissimilatory nitrate reduction processes in typical Chinese paddy soils: rates, relative contributions and influencing factors. Environ. Sci. Technol. 50, 9972–9980.
- She, Z.L., Zhao, L.T., Zhang, X.L., Jin, C.J., Guo, L., Yang, S.Y., et al., 2016. Partial nitrification and denitrification in a sequencing batch reactor treating high-salinity wastewater. Chem. Eng. J. 288, 207–215.
- Wang, D.Q., Chen, Z.L., Wang, J., Xu, S.Y., Yang, H.X., Chen, H., et al., 2007. Summertime denitrification and nitrous oxide exchange in the intertidal zone of the Yangtze Estuary. Estuar. Coast. Shelf. S. 73, 43–53.
- Welsh, D., Bartoli, M., Nizzoli, D., Castaldelli, G., Riou, S.A., Viaroli, P., 2000. Denitrification, nitrogen fixation, community primary productivity and inorganic-N and oxygen fluxes in an intertidal Zostera noltii meadow. Mar. Ecol. Prog. Ser. 208, 65–77.
- White, D., Howes, B., 1994. Long-term <sup>15</sup>N-nitrogen retention in the vegetated sediments of a New England salt marsh. Limnol. Oceanogr. 39, 1878–1892.
- Xia, X., Jia, Z., Liu, T., Zhang, S., Zhang, L., 2017. Coupled nitrification-denitrification caused by suspended sediment (sps) in rivers: importance of sps size and composition. Environ. Sci. Technol. 51, 212–221.
- Yabe, T., Ishii, Y., Amano, Y., Koga, T., Hayashi, S., Nohara, S., et al., 2009. Green tide formed by free-floating Ulva spp. at Yatsu tidal flat. Japan Limnology. 10, 239–245.
- Yang, F., Mi, T., Chen, H., Yao, Q., 2019. Developing numeric nutrient criteria for the Yangtze River Estuary and adjacent waters in China. J. Hydrol. doi:10.1016/ j.jhydrol.2019.124188.
- Yang, W.M., Zhao, Q., Lu, H., Ding, Z., Meng, L., Chen, G.H., 2015. Sulfide-driven autotrophic denitrification significantly reduces N<sub>2</sub>O emissions. Water. Res. 90, 176–184.
- Yin, G.Y., Hou, L.J., Liu, M., Liu, Z.F., Gardner, W.S., 2014. A novel membrane inlet mass spectrometer method to measure <sup>15</sup>NH<sub>4</sub><sup>+</sup> for isotope-enrichment experiments in aquatic ecosystems. Environ. Sci. Technol. 48, 9555–9562.
- Yin, G.Y., Hou, L.J., Liu, M., Zheng, Y.L., Li, X.F., Lin, X.B., et al., 2016. Effects of thiamphenicol on nitrate reduction and N<sub>2</sub>O release in estuarine and coastal sediments. Environ. Pollut. 214, 265–272.
- Yin, G.Y., Hou, L.J., Zong, H.B., Ding, P.X., Liu, M., Zhang, S.F., et al., 2015. Denitrification and Anaerobic ammonium oxidization across the sediment–water interface in the hypereutrophic ecosystem, Jinpu Bay, in the northeastern coast of China. Estuar. Coast. 38, 211–219.
- Zheng, Y.L., Hou, L.J., Chen, F.Y., Zhou, J., Liu, M., Yin, G.Y., et al., 2019. Denitrifying anaerobic methane oxidation in intertidal marsh soils: occurrence and environmental significance. Geoderma. doi:10.1016/j.geoderma.2019.113943.